

Lecture 22: Energy balance

Logistics: HW6 still missing one

→ next Tuesday 9:30am last chance

HW7 still missing two

Last time: local Eulerian balance laws

mass: $\dot{\rho} + \rho \nabla \cdot \underline{v} = 0$

$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \underline{v}) = 0$$

conservation
→ advection mass flux

lin. mom.: $\rho \dot{\underline{v}} - \nabla \cdot \underline{\underline{\underline{\sigma}}} = \rho \underline{b}$

$$\frac{\partial}{\partial t} (\rho \underline{v}) + \nabla \cdot (\rho \underline{v} \otimes \underline{v} - \underline{\underline{\underline{\sigma}}}) = \rho \underline{b}$$

ang. mom.: $\underline{\underline{\underline{\sigma}}} = \underline{\underline{\underline{\sigma}}}^T$

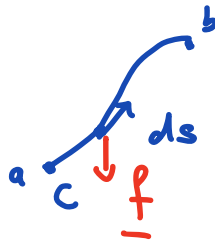
- Today:
- Review of thermodynamics
 - Work, heat, power, energy
 - local Eulerian balance
 - stress power
 - entropy

Work & Power

Work: energy transferred by application of force along a distance.

$$W = F s$$

$$W = \int_c \underline{f} \cdot \underline{ds} = \int_{t_1}^{t_2} \underline{f} \cdot \underbrace{\frac{d\underline{s}}{dt}}_{\underline{v}} dt$$



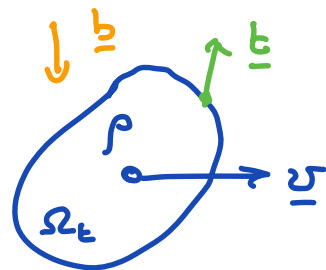
$$W = \int_{t_1}^{t_2} \underline{f} \cdot \underline{v} dt = \int \frac{dW}{dt} dt \Rightarrow \frac{dW}{dt} = \underline{f} \cdot \underline{v}$$

Power is the rate of work

$$\mathcal{P} = \frac{dW}{dt} = \underline{f} \cdot \underline{v}$$

Working rate of work being done \sim power

Exchange of mechanical energy between body and surrounding
Kinetic energy



$$\mathcal{K}[\Omega_t] = \int_{\Omega_t} \frac{1}{2} \rho(\underline{x}, t) |\underline{v}(\underline{x}, t)|^2 dV_x$$

Power of external forces on Ω_t

$$\begin{aligned}\mathcal{P}[\Omega_t] &= \underline{\tau}[\Omega_t] \cdot \underline{v} = \underline{\tau}_b[\Omega_t] \cdot \underline{v} + \underline{\tau}_s[\Omega_t] \cdot \underline{v} \\ &= \int_{\Omega_b} \rho \underline{b} \cdot \underline{v} dV_x + \int_{\partial\Omega_b} \underline{t} \cdot \underline{v} dA_x\end{aligned}$$

Net working $\mathcal{W}[\Omega_t]$ of external forces on Ω_t

$$\mathcal{W}[\Omega_t] = \mathcal{P}[\Omega_t] - \frac{d}{dt} \mathcal{K}[\Omega_t]$$

$\mathcal{W}[\Omega_t] > 0$: mechanical energy is stored in body

$\mathcal{W}[\Omega_t] < 0$: " " " " released from body

Temperature and heat

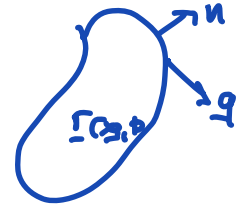
macroscopic quantity \rightarrow magnitude of microscopic
velocity fluctuations

absolute temp. field $\theta(\underline{x}, t) > 0$

Heat is energy associated with velocity fluctuations

⇒ Thermal energy

Heating is rate of heat gain/loss



I) body heating: $Q_b[\Omega_t] = \int_{\Omega_t} \rho r \, dV_x$

$r(x,t)$ = heat supply/loss per unit mass

II) surface heating: $Q_s[\Omega_t] = - \int_{\partial\Omega_t} \mathbf{q} \cdot \mathbf{n} \, dA_x$

$\mathbf{q}(x,t)$ = heat flux vector

Net heating

$$Q[\Omega_t] = Q_b[\Omega_t] + Q_s[\Omega_t] = \int_{\Omega_t} \rho r \, dV - \int_{\partial\Omega_t} \mathbf{q} \cdot \mathbf{n} \, dA$$

Internal energy and 1st law of thermodynamics

Internal energy is energy not associated with kinetic energy.

Internal energy = thermal (heat) + mechanical (elastic)

neglect electromagnetic & chemical energy

$$U[\Omega_t] = \int_{\Omega} \rho u \, dV_x$$

$u(x,t)$ is internal energy density per unit mass

First law of Thermo:

$$\frac{d}{dt} U[\Omega_t] = Q[\Omega_t] + W[\Omega_t]$$

↑ ↑

$$W[\Omega_t] = P[\Omega_t] - \frac{d}{dt} K[\Omega_t]$$

$$\Rightarrow \frac{d}{dt} \underbrace{(U[\Omega_t] + K[\Omega_t])}_{\text{total energy}} = Q[\Omega_t] + P[\Omega_t]$$

In some cases power of an external force can be written $P[\Omega_t] = - \frac{d}{dt} G[\Omega_t]$

$$\frac{d}{dt} (U[\Omega_t] + K[\Omega_t] + G[\Omega_t]) = Q[\Omega_t]$$

Balance of Energy in local Eulerian form

Net working in Eulerian form?

$$\text{Power: } \mathcal{P} = \underline{f} \cdot \underline{v}$$

$$\text{Newton's 2nd law: } \underline{f} = m \underline{a} = m \dot{\underline{v}}$$

$$\text{lin. mom. balance: } \rho \dot{\underline{v}} = \nabla \cdot \underline{\underline{\underline{\sigma}}} + \rho \underline{b}$$

dot product $(\rho \dot{\underline{v}}) \cdot \underline{v}$:

$$\rho \underline{v} \cdot \dot{\underline{v}} = (\nabla \cdot \underline{\underline{\underline{\sigma}}}) \cdot \underline{v} + \rho \underline{b} \cdot \underline{v}$$

integrate over Ω_t to identify \mathcal{K} , \mathcal{P}

$$\int_{\Omega_t} \rho \underline{v} \cdot \dot{\underline{v}} \, dV_x = \int_{\Omega_t} \underline{\underline{\underline{\sigma}}} \cdot \underline{v} + \rho \underline{b} \cdot \underline{v} \, dV$$

$$\text{use identity: } \nabla \cdot (\underline{\underline{\underline{A}}}^T \underline{b}) = \underline{\underline{\underline{A}}} \cdot \underline{\underline{\underline{\nabla}}} \underline{b} + \underline{\underline{\underline{A}}} : \underline{\underline{\underline{\nabla}}} \underline{b}$$

$$(\nabla \cdot \underline{\underline{\underline{\sigma}}}) \cdot \underline{v} = \nabla \cdot (\underline{\underline{\underline{\sigma}}}^T \underline{v}) - \underline{\underline{\underline{\sigma}}} : \underline{\underline{\underline{\nabla}}} \underline{v}$$

substitute

$$\int_{\Omega_t} \rho \underline{v} \cdot \dot{\underline{v}} \, dV_x = - \int_{\Omega_t} \underline{\underline{\underline{\sigma}}} : \underline{\underline{\underline{\nabla}}} \underline{v} + \nabla \cdot (\underline{\underline{\underline{\sigma}}} \underline{v}) + \rho \underline{b} \cdot \underline{v} \, dV_x$$

$$\text{Use property } \underline{\underline{\underline{S}}} : \underline{\underline{\underline{D}}} = \underline{\underline{\underline{S}}} : \text{sym}(\underline{\underline{\underline{D}}}) \text{ if } \underline{\underline{\underline{S}}} = \underline{\underline{\underline{S}}}^T$$

$$\underline{\underline{\underline{\sigma}}} = \underline{\underline{\underline{\sigma}}}^T \rightarrow \underline{\underline{\underline{\sigma}}} : \underline{\underline{\underline{\nabla}}} \underline{v} = \underline{\underline{\underline{\sigma}}} : \text{sym}(\underline{\underline{\underline{\nabla}}} \underline{v}) = \underline{\underline{\underline{\sigma}}} : \underline{\underline{\underline{d}}}$$

$$\underline{\underline{\underline{d}}} \doteq \frac{1}{2} (\underline{\underline{\underline{\nabla}}} \underline{v} + \underline{\underline{\underline{\nabla}}} \underline{v}^T) \text{ rate of def. tensor}$$

$$\int_{\Omega_t} \rho \underline{v} \cdot \underline{\dot{v}} \, dV_x = - \int_{\Omega_t} \underline{\underline{\underline{\sigma}}} : \underline{\underline{\underline{d}}} + \rho \underline{b} \cdot \underline{v} \, dV_x + \int_{\partial \Omega_t} \underline{\underline{\underline{\sigma}}} \underline{v} \cdot \underline{n} \, dA_x$$

definition of transpose: $\underline{\underline{\underline{\sigma}}} \underline{v} \cdot \underline{n} = \underline{v} \cdot \underline{\underline{\underline{\sigma}}}^T \underline{n}$
 $= \underline{v} \cdot \underline{\underline{\underline{\sigma}}} \underline{n} = \underline{v} \cdot \underline{t}$

$$\int_{\Omega_t} \rho \underline{v} \cdot \underline{\dot{v}} \, dV_x = - \int_{\Omega_t} \underline{\underline{\underline{\sigma}}} : \underline{\underline{\underline{d}}} \, dV_x + \underbrace{\int_{\Omega_t} \rho \underline{b} \cdot \underline{v} \, dV + \int_{\partial \Omega_t} \underline{t} \cdot \underline{v} \, dA_x}_{\text{rate of change of } \mathcal{P}[\Omega_t]}$$

Identify l.h.s. as kinetic energy

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$$\frac{d}{dt} \mathcal{K}[\Omega_t] = \frac{d}{dt} \int_{\Omega_t} \frac{1}{2} \rho \underline{v} \cdot \underline{v} \, dV_x = \frac{1}{2} \int_{\Omega_t} \rho \frac{d}{dt} (\underline{v} \cdot \underline{v}) \, dV_x$$

$$\frac{d}{dt} (v_i v_i) = \dot{v}_i v_i + v_i \dot{v}_i = 2 (v_i \dot{v}_i) = 2 \underline{v} \cdot \underline{\dot{v}}$$

substituting \mathcal{K} & \mathcal{P}

$$\frac{d}{dt} \mathcal{K}[\Omega_t] + \underbrace{\int_{\Omega_t} \underline{\underline{\underline{\sigma}}} : \underline{\underline{\underline{d}}} \, dV_x}_{\mathcal{W}[\Omega_t]} = \mathcal{P}[\Omega_t]$$

$$\dot{W}[\Omega_t] = \dot{P}[\Omega_t] - \frac{d}{dt} K[\Omega_t]$$

$$\dot{W}[\Omega_t] = \int \underline{\underline{\sigma}} : \underline{\underline{d}} dV$$

quantity $\underline{\underline{\sigma}} : \underline{\underline{d}}$ is called stress power associated with a motion and represents the ~~to~~ rate of work ~~of~~ by internal forces in a body.

Local Eulerian energy balance

Integral form of first law of thermodynamics

$$\frac{d}{dt} U[\Omega_t] = Q[\Omega_t] + \dot{W}[\Omega_t]$$

where $U[\Omega_t] = \int_{\Omega_t} \rho u dV_x$

$$Q = \int_{\Omega_t} \rho r dV_x - \int_{\partial\Omega_t} \underline{\underline{q}} \cdot \underline{\underline{n}} dA_x$$

$$\dot{W} = \int_{\Omega_t} \underline{\underline{\sigma}} : \underline{\underline{d}} dV_x$$

substituting

In simple case:

$$u = c_p T \quad c_p = \text{heat capacity (at const pressure)}$$

$$q = -\kappa \nabla T \quad \text{Fourier's law}$$

\uparrow thermal conductivity

substitute

$$\frac{\partial}{\partial t} (\rho c_p T) + \nabla \cdot (\underline{v} \rho c_p T - \kappa \nabla T) = \underline{\underline{\underline{\sigma}}} : \underline{\underline{\underline{d}}} + \rho r$$

assume $\rho, c_p, \kappa = \text{const.}$ simplify

$$\rho c_p \frac{\partial T}{\partial t} + \nabla \cdot (\underbrace{\underline{v} \rho c_p T}_{\text{adv}} - \underbrace{\kappa \nabla T}_{\text{cond}}) = \underline{\underline{\underline{\sigma}}} : \underline{\underline{\underline{d}}} + \rho r$$

$d = \frac{1}{2}(\nabla v + \nabla v^T)$

advection - conduction eqn

$$\underline{v} = \underline{0}$$

\Rightarrow heat equation

$$\rho c_p \frac{\partial T}{\partial t} = \kappa \underbrace{\nabla \cdot \nabla T}_{\nabla^2} + \rho r$$

$$\frac{\partial T}{\partial t} = \underbrace{\frac{\kappa}{\rho c_p}}_{\text{diffusion coefficient}} \nabla^2 T + \frac{r}{c_p}$$

α = thermal diffusivity

$$\frac{\partial T}{\partial t} = \alpha \nabla^2 T + \frac{\Gamma}{c_p}$$

Heat equation